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SCI304BX Honors Chemistry Semester 2 Unit 2: Solutions K12 Curriculum Lawrence Virtual High School. Terms in this set (18) saturated. a solution in which the greatest possible amount of solute is dissolved. solute. a substance that dissolves in solvent. solution.

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The lamp will be lit when its electrodes are immersed into the salt solution. Sugar dissolution in water: $C_6H_{12}O_6(s) + H_2O(l) \rightarrow [C_6H_{12}O_6](aq)$ [No ions are produced] The sugar solution does not produce any ions and hence no electricity is conducted and the lamp will not be lit.

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A solution formed by dissolving a metal in a metal or a non-metal in a metal while the components are molten (melted) is called a(n) _____.
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2.15. An aqueous solution of 2 percent non-volatile solute exerts a pressure of 1.004 bar at the boiling point of the solvent. What is the molecular mass of the solute? Sol: According to Raoult's Law, 2.16 Heptane and octane form an ideal solution. At 373 K, the vapour pressures of the two liquid components are 105.2 kPa and 46.8 kPa ...

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